

# Crystal Growth, Structure and Properties of a New Ternary Intermetallic Compound $\text{Nb}_5\text{Sn}_2\text{Ga}$

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**Abstract:** Single crystals of a new ternary intermetallic compound  $\text{Nb}_5\text{Sn}_2\text{Ga}$  were synthesized by the high-temperature Nb–Sn–Ga system solutions under a He gas. The Nb and Ga were mixed together at atomic ratios of between 1.5:1 and 6:1. Tin was added to these mixtures at a ratio of 5:1 in weight. The mixture was heated at 1400°C for 10 h, and then slowly cooled down at a rate of 1°C h<sup>-1</sup> to 1000°C. Then the furnace was rapidly cooled down to room temperature. Single crystals of  $\text{Nb}_5\text{Sn}_2\text{Ga}$ , having a metallic luster, were generally obtained in the form of prismatic shape extending in the [001] direction and with (100) and (110) faces. The largest crystals prepared have maximum dimensions of about 10 mm × 1 mm × 1 mm. The nominal compositions of  $\text{Nb}_5\text{Sn}_2\text{Ga}$  crystals were widely changed as Nb to Ga 2:1, 3:1 and 4:1, and the compositions of the obtained crystals were close to the  $\text{Nb}_5\text{Sn}_2\text{Ga}$ . Single crystal X-ray analysis showed that the new compound  $\text{Nb}_5\text{Sn}_2\text{Ga}$  has a tetragonal symmetry with  $a=1.0586(2)\text{nm}$ ,  $c=0.5177(1)\text{nm}$ , space group  $I4/mcm$ , ordered  $\text{W}_5\text{Si}_3$ -type structure. The superconductivity, electric property, Vickers microhardness and oxidation resistivity on the single crystals of  $\text{Nb}_5\text{Sn}_2\text{Ga}$  were determined. The compound shows superconductivity at  $T_c=1.75\text{ K}$  and  $\Delta T_c=140\text{ mK}$ . The electrical resistivity determined on crystal is about  $120 \times 10^{-6}\ \Omega\text{cm}$  at room temperature. The Vickers microhardness value on (001) plane and (100) or (110) planes of crystals is in the range of  $H_v=8.5\text{--}10.1\text{ GPa}$ . The oxidation of  $\text{Nb}_5\text{Sn}_2\text{Ga}$  crystal starts at about 562°C. The final oxidation products were  $\text{NbO}_2$  (monoclinic),  $\text{Nb}_{12}\text{O}_{29}$  (orthorhombic),  $\text{Nb}_2\text{O}_5$  (monoclinic),  $\text{SnO}_2$  (tetragonal) and  $\text{Ga}_2\text{O}_3$  (rhombohedral), respectively.

**Keywords:**  $\text{Nb}_5\text{Sn}_2\text{Ga}$ , Single crystal, Crystal morphology, Crystal structure, Superconductivity, Vickers microhardness, Thermal properties

## 1. Introduction

Single crystals of a new superconducting ternary intermetallic compound  $\text{Nb}_5\text{Sn}_2\text{Ga}$  isostructural with tetragonal  $\text{Nb}_5\text{Sn}_2\text{Si}$  (space group  $I4/mcm$ ) [1] were pre-

pared from self-component tin solution [2, 3]. It was reported that the crystal growth and crystal structure [2], the results of X-ray photoelectron spectroscopy (XPS) (the chemical state of Nb, Sn and Ga) [4], and the superconductivity [5] and electric property [6] on as-grown  $\text{Nb}_5\text{Sn}_2\text{Ga}$  crystals. However, there is a little information about chemical and physical properties of  $\text{Nb}_5\text{Sn}_2\text{Ga}$  crystal. The purpose of this paper is to clarify the conditions for growing relatively large crystals of  $\text{Nb}_5\text{Sn}_2\text{Ga}$  in high temperature tin solution by a self flux. X-ray diffraction for  $\text{Nb}_5\text{Sn}_2\text{Ga}$  single crystals grown from three different compositions were carried out. The crystal size, crystal morphology, crystallographic data, superconductivity, electrical resistivity and Vickers microhardness of the compounds were determined, and the oxidation reaction heated in air was studied by thermogravimetric (TG) analysis and differential thermal analysis (DTA). The oxidation products were analyzed by a X-ray diffractometer (XRD) at room temperature.

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**Table 1.** Typical growth conditions and chemical formula of Nb<sub>5</sub>Sn<sub>2</sub>Ga crystals.

Run No.	Composition of the starting mixture in atomic ratio Nb : Ga	Chemical formula of crystals	Crystal shape
1	1.5 : 1	—	Powder
2	2 : 1	Nb <sub>5</sub> Sn <sub>1.95</sub> Ga <sub>1.08</sub>	Rectangular, powder
3	3 : 1	Nb <sub>5</sub> Sn <sub>1.97</sub> Ga <sub>1.06</sub>	Rectangular, powder
4	4 : 1	Nb <sub>5</sub> Sn <sub>2.12</sub> Ga <sub>0.94</sub>	Rectangular, powder
5	6 : 1	—	Powder

## 2. Experimental details

### 2.1 Preparation of crystals

Synthesis of the ternary Nb<sub>5</sub>Sn<sub>2</sub>Ga crystals was performed by the high-temperature solution growth method using tin as a self flux. The starting materials used were small pieces of 99.9% Nb, 99.999% Ga button and 99.999% Sn powder. The Nb and Ga were mixed together at atomic ratios of between 1.5:1 and 6:1 (Table 1). Tin was added to these mixtures at a ratio of 5:1 in weight. The mixture of starting materials was placed in a high purity (99.9%) dense alumina crucible. A pure helium gas flow at a rate of 200 ml min<sup>-1</sup> was introduced into the furnace as a protecting atmosphere against oxidation. The temperature of the furnace was raised at a rate of 400°C h<sup>-1</sup> up to 1400°C and kept for 10 h, and then slowly cooled down at a rate of 1°C h<sup>-1</sup> to 1000°C. Then the furnace was rapidly cooled down to room temperature. The crystals were separated by dissolving Sn in dilute HCl (6 M) for 2 to 4 days. As shown in Table 1, rectangular single crystals of Nb<sub>5</sub>Sn<sub>2</sub>Ga were obtained when the ratio of Nb to Ga were taken as 2:1, 3:1 and 4:1.

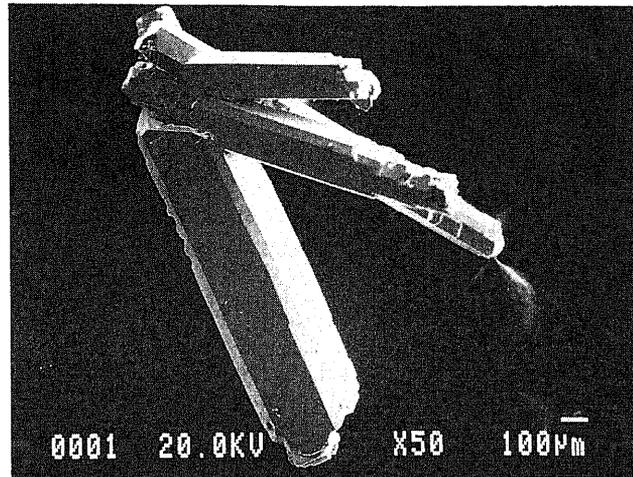
### 2.2 Chemical and X-ray diffraction analyses

Relatively large crystals of Nb<sub>5</sub>Sn<sub>2</sub>Ga were selected under stereomicroscope, and chemical compositions of the crystals grown were determined by means of the inductively coupled plasma (ICP) technique. The crystal morphology was examined by an optical microscope and a scanning electron microscope (SEM). Some of the crystals were examined with an energy-dispersive detector (EDX). X-ray diffraction analyses were performed using a Burger precession camera with zirconium-filtered Mo K $\alpha$  radiation and a four-circle X-ray diffractometer with graphite-monochromatized Mo K $\alpha$  radiation.

### 2.3 Properties

#### 2.3.1 Measurement of hardness, superconductivity, electric and oxidation resistivity

The Vickers microhardness of the as-grown Nb<sub>5</sub>Sn<sub>2</sub>Ga crystal was measured in several directions on (001) plane and (100) or (110) planes, at room temperature in air. A load of 0.98 N was applied for 15 sec at about 5 to 9 points for each crystal, and the values obtained were averaged. The superconducting transition temperature  $T_c$  was determined by measuring the AC susceptibility. The electrical resistivity of the crystals was measured by the



**Fig. 1** A SEM photograph of Nb<sub>5</sub>Sn<sub>2</sub>Ga (Run No. 3) single crystals.

direct-current four-probe technique between 1.5 and 298 K. Thermogravimetric (TG) analysis and differential thermal analysis (DTA) were performed up to 1200°C to study the oxidation of crystals in air. Specimens of 15–25 mg were heated at rate of 10°C h<sup>-1</sup>.

## 3. Results and discussion

### 3.1 Growth, morphology and structure of the single crystals

Rectangular single crystals were obtained when the ratio of Nb to Ga in the melt were taken as 2:1, 3:1 and 4:1 (Table 1). The ternary single crystals Nb<sub>5</sub>Sn<sub>2</sub>Ga were generally obtained in the form of prismatic shape extending in the [001] direction, and with (001) and (110) faces. These crystals had a silver metallic luster. The maximum sizes of the crystals were about 10 mm × 1 mm × 1 mm (Fig. 1). The range of solid solution of the compound is very narrow. The single crystals obtained from run number 3 were closest to the stoichiometric composition. Therefore, the Vickers microhardness, superconductivity, electrical resistivity and oxidation were measured for the sample of run number 3. Also, precession photographs taken along the [001] and [110] zone-axes using the crystal obtained from run 3 are shown in Fig. 2. The results of crystal structure analyses for the compounds of runs 2–4 revealed that

Table 2. Crystal data and result of chemical analysis for Nb<sub>5</sub>Sn<sub>2</sub>Ga.

	Run No. 2	Run No. 3	Run No. 4
Formula	Nb <sub>5</sub> Sn <sub>2</sub> Ga	Nb <sub>5</sub> Sn <sub>2</sub> Ga	Nb <sub>5</sub> Sn <sub>2</sub> Ga
Crystal system	tetragonal	tetragonal	tetragonal
<i>a</i> (nm)	1.0584(1)	1.0586(2)	1.0606(2)
<i>c</i> (nm)	0.5175(1)	0.5177(1)	0.5196(1)
<i>V</i> ( $\times 10^{-3}$ nm <sup>3</sup> )	579.7	580.2	584.5
Space group	I4/mcm	I4/mcm	I4/mcm
Z	4	4	4
Chemical composition <sup>a)</sup>	Nb <sub>5</sub> Sn <sub>1.95</sub> Ga <sub>1.08</sub>	Nb <sub>5</sub> Sn <sub>1.97</sub> Ga <sub>1.06</sub>	Nb <sub>5</sub> Sn <sub>2.12</sub> Ga <sub>0.94</sub>
<i>D<sub>x</sub></i> (gcm <sup>-3</sup> )	8.84	8.83	8.77
<i>D<sub>m</sub></i> (gcm <sup>-3</sup> ) <sup>b)</sup>	—	8.83	—

<sup>a)</sup> Chemical analysis was done using the ICP method.

<sup>b)</sup> *D<sub>m</sub>* was determined by means of a pycnometer using H<sub>2</sub>O at 20°C.

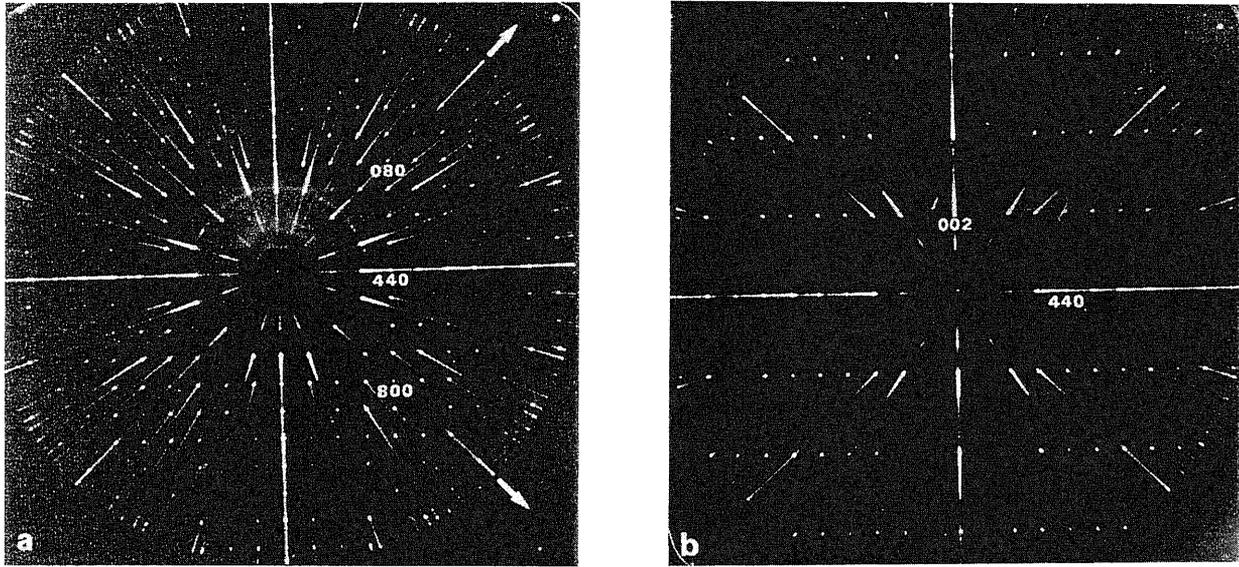


Fig. 2 X-ray precession photographs of Nb<sub>5</sub>Sn<sub>2</sub>Ga (Run No. 3): (a) [001] and (b) [110] zone axes.

they are all tetragonal, space group *I4/mcm*, and isostructural with Nb<sub>5</sub>Sn<sub>2</sub>Si [7], Ta<sub>5</sub>SnGa<sub>2</sub> [8, 9] and Nb<sub>5</sub>(Ge<sub>x</sub>, Sn<sub>1-x</sub>)<sub>2</sub>Ge (*x*=0.25) [10] whose basic structure is W<sub>5</sub>Si<sub>3</sub> type [1]. The unit cell parameters are listed in Table 2. The unit cell volume varies from 579.7 to 584.5 ( $\times 10^{-3}$ )nm<sup>3</sup> with increasing tin content in the chemical formula. According to ref. 11 the atomic radius (12-fold coordination) is 1.53 Å and 1.58 Å for Ga and for Sn, respectively. When the tin content is increased above its stoichiometric composition, the interatomic distance of Nb-Ga is also increased. This is caused by the partly replacement of Ga by Sn [5]. The arrangement of atoms in the crystal is shown in Fig. 3. The crystal structure of Nb<sub>5</sub>Sn<sub>2</sub>Ga is composed of two kinds of atomic polyhedra, GaNb(1)<sub>8</sub> and Nb(2)Sn<sub>4</sub>. Both polyhedra form columns parallel to the *c*-axis, respectively.

Table 3. Superconductivity and electric properties of Nb<sub>5</sub>Sn<sub>2</sub>Ga crystals.

<i>T<sub>c</sub></i>	1.75 K (midpoint)
$\Delta T_c$	140 mK
$\rho$ (298 K)	$120 \times 10^{-6}$ Ω cm
$\rho$ (4.2 K)	$10 \times 10^{-6}$ Ω cm

### 3.2 Superconductivity and electric property

The real part of the susceptibility as a function of the temperature is shown in Fig. 4. *T<sub>c</sub>* was determined to be 1.75 K at the midpoint of the transition, and  $\Delta T_c$  defined by 10 and 90% of the transition was 140 mK. The electrical resistivity of as-grown Nb<sub>5</sub>Sn<sub>2</sub>Ga crystals was meas-

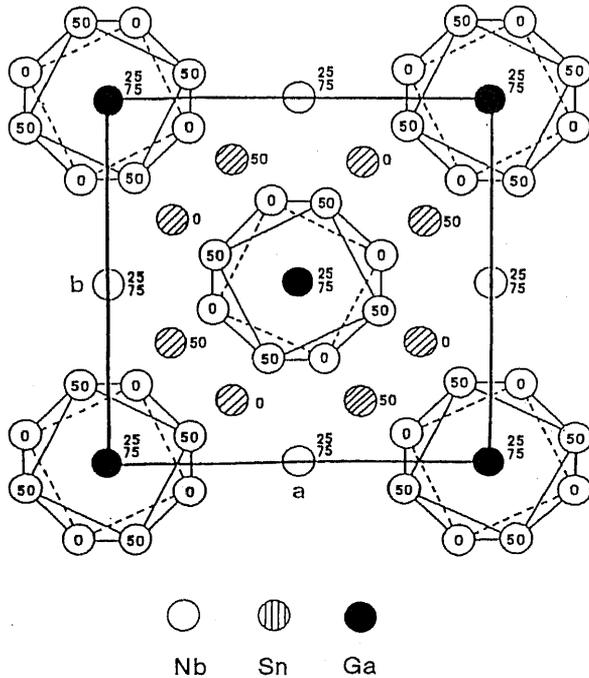


Fig. 3 A view of the crystal structure for  $\text{Nb}_5\text{Sn}_2\text{Ga}$  along the  $c$ -axis. Open circles are Nb, solid circles are Ga and shaded circles are Sn atoms, respectively.

Table 4. Vickers microhardness of  $\text{Nb}_5\text{Sn}_2\text{Ga}$  crystals.

Compound	Indentation plane	Microhardness Hv (GPa)
$\text{Nb}_5\text{Sn}_2\text{Ga}$	(001)	10.1–9.1
	(100) or (110)	8.9–8.5

ured at room temperature on the (100) or (110) planes. The superconductivity and electrical resistivity are listed in Table 3. The resistivity is about  $120 \times 10^{-6} \Omega\text{cm}$  and is metallic. The electrical resistivity value of the  $\text{Nb}_5\text{Sn}_2\text{Ga}$  crystals was found to be larger than the value of  $\text{Nb}_5\text{Sn}_2\text{Si}$ ,  $\text{Ta}_5\text{Ga}_2\text{Sn}$ , and  $\text{Nb}_5(\text{Ge}_x\text{Sn}_{1-x})_2\text{Ge}$  crystals with ordered- $\text{W}_5\text{Si}_3$  type structure [10].

### 3.3 Microhardness

The Vickers microhardness [12] of as-grown  $\text{Nb}_5\text{Sn}_2\text{Ga}$  were listed in Table 4. The microhardness values as measured on the (100) or (110) face of  $\text{Nb}_5\text{Sn}_2\text{Ga}$  crystals are found to be relatively lower than the values of (001) face. This anisotropic nature of microhardness seems to be related to the difference in the number of Nb atoms per unit area between the (001) plane and the (100) or (110) planes [5]. The greater number of Nb atoms on the (001) plane compared with (100) or (110) planes gave a higher value of microhardness.

### 3.4. Oxidation reaction

The oxidation reaction was studied by TG and DTA [13, 14], as shown in Fig. 5. The TG curve show that oxida-

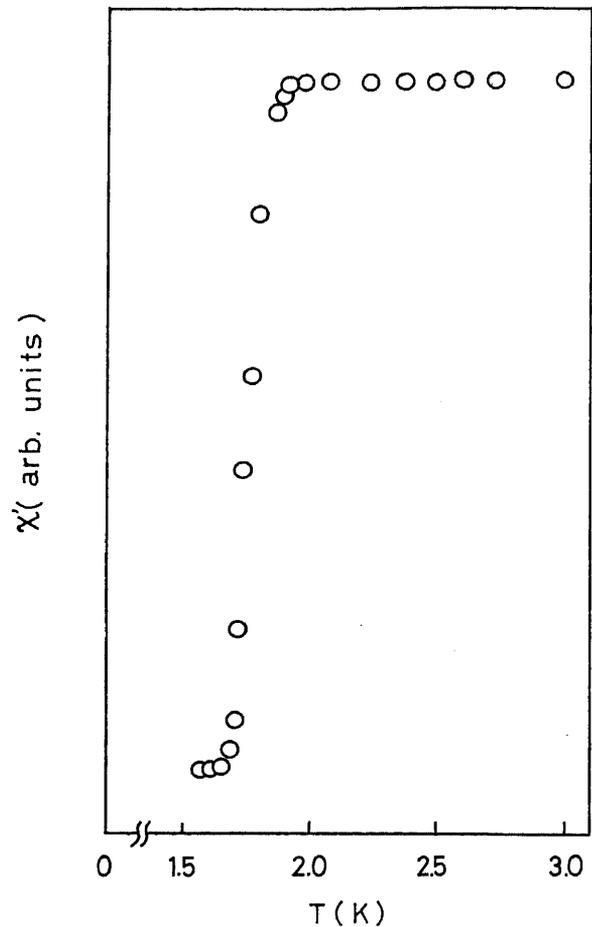


Fig. 4 Temperature dependence of the AC magnetic susceptibility of  $\text{Nb}_5\text{Sn}_2\text{Ga}$ .

tion reaction starts at  $562^\circ\text{C}$  for  $\text{Nb}_5\text{Sn}_2\text{Ga}$ . Weight gain of the specimen after TG determination, heated up to  $1200^\circ\text{C}$  in air, was measured at about 37.3 wt%. The exothermic peaks of the DTA curve were found at about  $608^\circ\text{C}$ ,  $796^\circ\text{C}$  and  $925^\circ\text{C}$ , respectively. The oxidation products were analyzed by a XRD at room temperature. Fig. 6 shows the XRD pattern after the crystal was heated at  $1200^\circ\text{C}$  for 5 min, and the oxidation products were  $\text{NbO}_2$  (monoclinic, space group  $-$ ),  $\text{Nb}_{12}\text{O}_{29}$  (orthorhombic, space group  $Amma$ ),  $\text{Nb}_2\text{O}_5$  (monoclinic, space group  $P2$ ),  $\text{SnO}_2$  (tetragonal, space group  $P4_2/mnm$ ), and  $\text{Ga}_2\text{O}_3$  (rhombohedral, space group  $R\bar{3}c$ ), respectively.

## 4. Conclusions

The purpose of this study is to clarify the conditions for growing relatively large crystals of  $\text{Nb}_5\text{Sn}_2\text{Ga}$  in high temperature tin solution by a self flux under a He gas. The mixture of the raw materials was raised at a rate of  $400^\circ\text{C}^{-1}$  up to  $1400^\circ\text{C}$  and kept for 10 h, and then slowly cooled down to room temperature. X-ray diffraction of  $\text{Nb}_5\text{Sn}_2\text{Ga}$  single crystals grown from three different com-

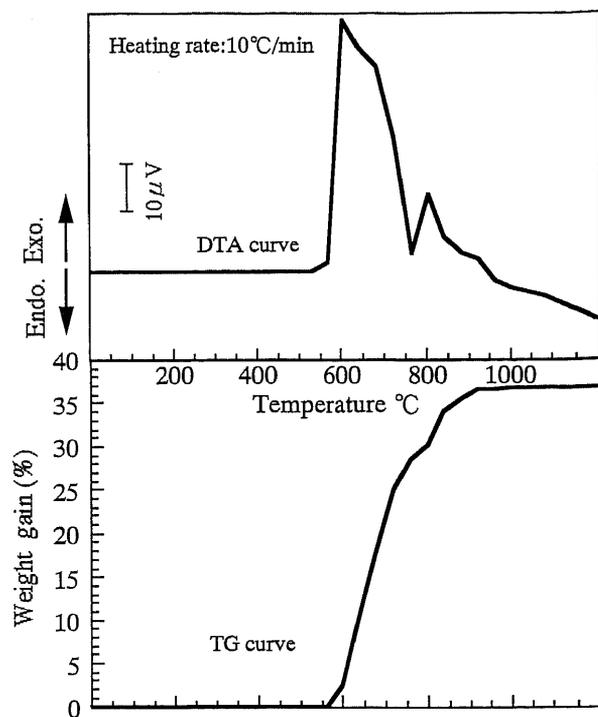


Fig. 5 Differential thermal analysis (DTA) and thermal gravimetric (TG) curves for Nb<sub>5</sub>Sn<sub>2</sub>Ga single crystals. Samples were heated in air at a rate of 10°C/h<sup>-1</sup>.

positions were carried out. The crystal size, crystal morphology, crystallographic data, superconductivity, electrical resistivity and Vickers microhardness of the com-

pounds were measured, and the oxidation reaction heated in air was studied. The conclusions derived from this study are as follows:

(1) Single crystals of Nb<sub>5</sub>Sn<sub>2</sub>Ga were generally obtained in the form of prismatic shape extending in the [001] direction and with (100) and (110) faces.

(2) The largest crystals prepared have maximum dimensions of about 10 mm × 1 mm × 1 mm.

(3) The nominal compositions of Nb<sub>5</sub>Sn<sub>2</sub>Ga crystals were widely changed as Nb to Ga 2:1, 3:1 and 4:1.

(4) Nb<sub>5</sub>Sn<sub>2</sub>Ga has a tetragonal symmetry with  $a=1.0586(2)\text{nm}$ ,  $c=0.5177(1)\text{nm}$ , space group  $I4/mcm$  and ordered W<sub>5</sub>Si<sub>3</sub>-type structure.

(5) The compound shows superconductivity at  $T_c=1.75\text{K}$  and  $\Delta T_c=140\text{mK}$ . The electrical resistivity is about  $120 \times 10^{-6}\ \Omega\text{cm}$ .

(6) The Vickers microhardness value on (001) plane and (100) or (110) planes of crystals is in the range of  $H_v=8.5\text{--}10.1\text{GPa}$ .

(7) The oxidation of Nb<sub>5</sub>Sn<sub>2</sub>Ga crystal starts at about 562°C, and the final oxidation products were NbO<sub>2</sub> (monoclinic, space group  $-$ ), Nb<sub>12</sub>O<sub>29</sub> (orthorhombic, space group *Amma*), Nb<sub>2</sub>O<sub>5</sub> (monoclinic, space group *P2*), SnO<sub>2</sub> (tetragonal, space group *P4<sub>2</sub>/mmm*), and Ga<sub>2</sub>O<sub>3</sub> (rhombohedral, space group *R3c*).

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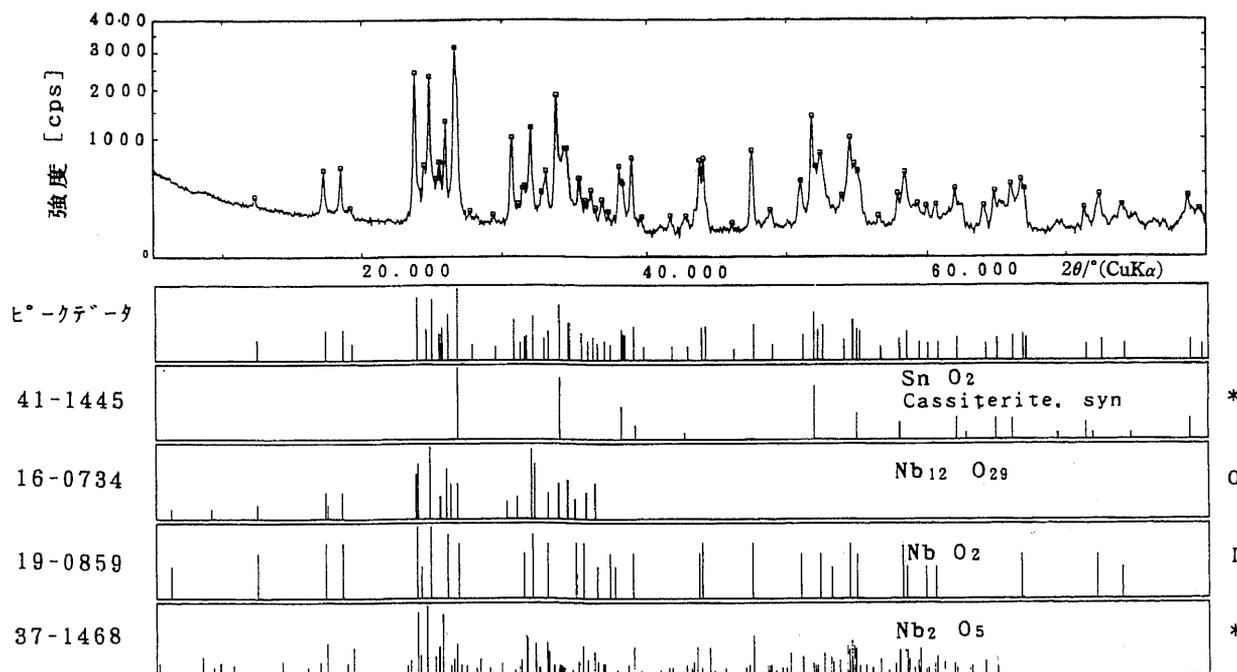


Fig. 6 XRD pattern of after the oxidation reaction. Nb<sub>5</sub>Sn<sub>2</sub>Ga crystal was heated at 1200°C for 5 min.

periments.

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